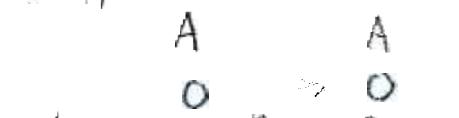


How to balance Equations

Steps

1 **Inspection** Just look at
If not
Wrote down all the elements
for the products + reactants
(Keep the order both sides)



Start The left if

The # of Atoms are smaller
Change the coefficient #)
of the 6 compound

Then change the # of atoms
below the equation.

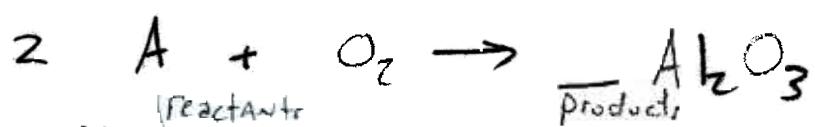
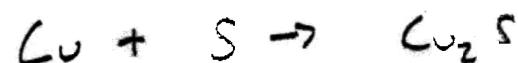
*If is compound
change both elements

Move to the right changing
the coeffic to

Go over + change the
products

You Are Balanced when
1 side th other

Check to see if the
coeffic to can be
educed



Al less than here

we change the coefficient to

2 The change the #

Atom below

